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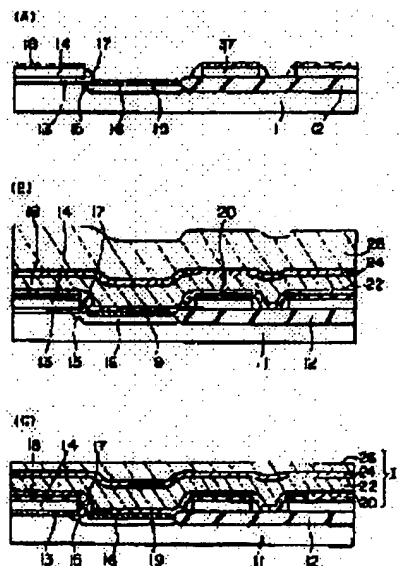
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(54) SEMICONDUCTOR DEVICE AND MANUFACTURE THEREOF

(57)Abstract:

PROBLEM TO BE SOLVED: To provide a semiconductor device comprising an interlayer insulating film which can be formed at a low temperature in comparison with a conventional interlayer insulating film using a BPSG film, is excellent in flatness and enables the formation of a highly reliable contact structure on a semiconductor substrate, and a method of manufacturing the device.

SOLUTION: A process of forming an interlayer insulating film 11 comprises a process wherein a hydrogen-containing silicon compound and hydrogen peroxide are reacted with each other by a chemical vapor deposition method to form a first silicon oxide film 22, a process wherein at least one kind of the compound or the element among a silicon compound, oxygen and an oxygen-containing compound and an impurities-containing compound are reacted with each other by the chemical vapor deposition method to form a second porous silicon oxide film 24, and a process wherein an annealing treatment is performed at a temperature of 600 to 850°C and the films 22 and 24 are microscopically formed. The film 22 is formed at a low temperature in comparison with a BPSG film and has excellent self-flattening characteristics in itself.



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DETAILED DESCRIPTION

[Detailed Description of the Invention]

[0001]

[The technical field to which invention belongs] Especially this invention relates to the semiconductor device which detailed-izing below a half micron is possible, and has a layer insulation film, and its manufacture method about a semiconductor device and its manufacture method.

[0002]

[Background of the Invention] In semiconductor devices, such as LSI, the bottom of the low temperature of the membrane formation temperature of a layer insulation film and flattening, and the formation technology of metal wiring have been an important technical problem with detailed-izing of an element, densification, and multilayering.

[0003] On the substrate in which the element was formed, a layer insulation film grows up a silicon oxide by the chemical-vapor-deposition method at low temperature first, after that, carries out gaseous phase reaction of the gas containing impurities, such as a silane compound, oxygen or ozone, and Lynn or boron, and forms a BPSG film by the thickness of hundreds of nm - about 1 micrometer. Then, by the so-called elevated-temperature flow which anneals at an elevated temperature in nitrogen-gas-atmosphere mind, a BPSG film is made to fluidize and the flattening is performed. Thus, after forming a through hole (contact hole) in the formed layer insulation film and forming the barrier layer which consists of titanium or titanium nitride, a metal wiring layer is formed.

[0004] Flattening of the layer insulation film using such a BPSG film is performed using the elevated-temperature flow property of a BPSG film, and flattening progresses, so that the high impurity concentration and the annealing temperature in a BPSG film are high. And in order to obtain flat nature and compactness with a sufficient BPSG film, it is required that an annealing temperature should be 850 degrees C or more.

[0005] However, in order to prevent generating of the punch through of an MOS transistor which turned minutely, it is important to suppress the breadth of the source with superfluous annealing **** and a drain impurity layer, for example, for that to process below 850 degrees C is desired. Moreover, in forming silicide layers, such as titanium, in the front face of the source which constitutes an MOS transistor, and a drain impurity layer, in elevated-temperature annealing, the field of a silicide layer is expanded more than required, and has become the factor which degrades a junction property. Since it is such, development of the technology which forms a layer insulation film at low temperature is demanded.

[0006]

[Problem(s) to be Solved by the Invention] Compared with the layer insulation film using the conventional BPSG film, membrane formation at low temperature is possible for the purpose of this invention, and it is to offer the semiconductor device which is excellent in flat nature and contains the layer insulation film on a semiconductor substrate which can form reliable contact structure, and its manufacture method.

[0007]

[Means for Solving the Problem] The process which forms a layer insulation film on the semiconductor substrate in which the manufacture method of the semiconductor device of this invention contains an element, The process which forms a barrier layer in the front face of the process which forms a through hole in the aforementioned layer insulation film, the aforementioned layer insulation film, and the aforementioned through hole, And the process which forms the aforementioned layer insulation film in the front face of the aforementioned barrier layer including the process which forms an electric conduction film is characterized by including following process (a) - (c) at least.

[0008] (a) The process which the compound containing at least one sort and impurity of the compound containing the process which a silicon compound and a hydrogen peroxide are made to react by the chemical-vapor-deposition method, and forms the 1st silicon oxide, the (b) silicon compound, oxygen, and oxygen is made to react by the chemical-vapor-deposition method, and forms the 2nd porous silicon oxide, and the process which performs annealing processing at (c) 600-850 degree C temperature.

[0009] According to the manufacture method of this semiconductor device, the layer which was excellent in flat nature can be formed by making a silicon compound and a hydrogen peroxide react by the chemical-vapor-deposition method, and forming the 1st silicon oxide according to a process (a). That is, the 1st silicon oxide formed at this process (a) has the self-flattening property of having a high fluidity and having excelled in itself. If the mechanism makes a silicon compound and a hydrogen peroxide react by the chemical-vapor-deposition method, it will be considered to be because for a silanol to be formed into a gaseous phase, and for a fluid good film to be formed when this silanol deposits on a wafer front face.

[0010] For example, when a mono silane is used as a silicon compound, a silanol is formed at the reaction shown by the following

formula (1), (1)', etc.

[0011] Formula (1)

$\text{SiH}_4 + 2\text{H}_2\text{O}_2 \rightarrow \text{Si}(\text{OH})_4 + 2\text{H}_2$ formula (1) The silanol formed by $\text{SiH}_4 + 3\text{H}_2\text{O}_2 \rightarrow \text{Si}(\text{OH})_4 + 2\text{H}_2\text{O} + \text{H}_2$ and the formula (1), and (1)' serves as a silicon oxide, when water ****s at the polycondensation reaction shown by the following formula (2).

[0012] Formula (2)

$\text{Si}(\text{OH})_4 \rightarrow$ as the $\text{SiO}_2 + 2\text{H}_2\text{O}$ aforementioned silicon compound -- a mono silane, a disilane, and SiH_2 -- organic silane compounds, such as inorganic silane compounds, such as Cl_2 , SiF_4 , and CH_3SiH_3 , and a TORIPURO pill silane, and a tetrapod ethoxy silane, etc. can be illustrated

[0013] Moreover, under 0-20-degree C temperature conditions, when the aforementioned silicon compound is an inorganic silicon compound, when the aforementioned silicon compound is an organic silicon compound, it is desirable [the membrane formation process of the aforementioned process (a)] to be carried out by the reduced pressure chemical-vapor-deposition method under 100-150-degree C temperature conditions. If temperature is higher than the aforementioned upper limit, when the polycondensation reaction of the aforementioned formula (2) progresses too much, the 1st silicon oxidization membrane fluidity will become low, and good flat nature will be hard to be obtained at this membrane formation process. Moreover, there is un-arranging [to which adsorption of a low and the decomposition moisture within a chamber and dew condensation out of a chamber occur, and it becomes difficult from the aforementioned lower limit for membrane formation equipment to control temperature].

[0014] As for the 1st silicon oxide formed at the aforementioned process (a), it is desirable to be formed by the thickness of the grade which can fully cover the level difference on the front face of a silicon substrate. The thickness of the 1st silicon oxide is 300-1000nm preferably, although the lower limit is dependent on the height of the irregularity on the front face of a silicon substrate containing an element. When the thickness of the 1st silicon oxide exceeds the aforementioned upper limit, a crack may be produced for the stress of the film itself.

[0015] At the aforementioned process (b), the compound containing at least one sort and impurity of the compound containing a silicon compound, oxygen, and oxygen is made to react by the chemical-vapor-deposition method, and the 2nd porous silicon oxide is formed on the 1st silicon oxide of the above.

[0016] This 2nd silicon oxide is porosity and it not only functions as a cap layer, but can emit gradually outside the gas constituents generated from the 1st silicon oxide in annealing processing of a next process (c). Furthermore, in addition to being porosity, impurities, such as Lynn and boron, and by adding Lynn preferably, this 2nd silicon oxide can ease the stress of this film by weakening the bonding strength between Si-O molecules of the silicon oxide which constitutes this film, and can constitute the layer which cannot break soft further easily moderately so to speak on this film. Moreover, there is a function as a getter of a movable ion in which impurities, such as Lynn included in this silicon oxide, have a bad influence on the reliance property of elements, such as alkali ion, as an important role of the 2nd silicon oxide of the above. The concentration of the impurity contained in the 2nd silicon oxide is 1 - 6 % of the weight preferably, when the point of the stress relaxation of the gettering function mentioned above or a film is taken into consideration.

[0017] Moreover, since the 2nd silicon oxide has the compression stress of 100-600MPa, in case the 1st silicon oxide carries out a polycondensation, it has the function to prevent that **** stress increases and a crack enters. Furthermore, the 2nd silicon oxide also has the function to prevent moisture absorption of the 1st silicon oxide.

[0018] As for the aforementioned process (b), it is desirable to be carried out under 300-450-degree C temperature conditions by the plasma-chemistry vapor growth by the RF 1MHz or less. By forming membranes on this temperature condition, it becomes easy to escape by the annealing initial stage by annealing of a process (c) from gas constituents, and the reliability of a device improves.

[0019] Moreover, as for the compound containing oxygen used at the aforementioned process (b), it is desirable that it is a dinitrogen oxide (N_2O). Since the dinitrogen oxide of the plasma state tends to react by using a dinitrogen oxide as reactant gas with the hydrogen bond (-H) of the silicon compound which constitutes the 1st silicon oxide, while forming the 2nd silicon oxide, desorption of the gasification component (hydrogen, water) of the 1st silicon oxide can be promoted.

[0020] The aforementioned process (b) may be performed by the ordinary-pressure chemical-vapor-deposition method under 300-550-degree C temperature conditions instead of a plasma-chemistry vapor growth. In this case, as for the compound containing the aforementioned oxygen used at the aforementioned process (b), it is desirable that it is ozone.

[0021] Furthermore, before forming the 2nd silicon oxide of the above at the aforementioned process (b), it is desirable to expose the 1st silicon oxide of the above to ozone atmosphere. Since ozone tends to react by passing through this process with the hydrogen bond (-H) of a silicon compound and the hydroxyl group (-OH) which constitute the 1st silicon oxide, the hydrogen in the 1st silicon oxide and desorption of water can be promoted.

[0022] Moreover, the thickness of the 2nd silicon oxide is 100nm or more preferably, when the point of prevention of flat nature and a crack is taken into consideration.

[0023] At the aforementioned process (c), by performing annealing processing at the temperature of 600-850 degrees C, the 1st and 2nd silicon oxides formed by the aforementioned process (a) and (b) turn precisely, and its moisture resistance improves in an insulating row.

[0024] That is, if it sees about the 1st silicon oxide, in the early stages of this annealing processing, the polycondensation reaction by the formula (2) mentioned above will be completed, water and hydrogen which are produced with this reaction will be emitted outside through the hole of the 2nd silicon oxide, and the 1st silicon oxide will be precisely formed, where a gasification component is fully removed. Moreover, the 2nd silicon oxide turns into a precise film from porosity by annealing processing.

[0025] In this annealing processing, while being able to make precise enough the 1st and 2nd silicon oxides by making temperature into 600 degrees C or more, the impurity of the source which constitutes the MOS device, for example, and a drain diffusion layer is fully activable. Moreover, rather than the temperature needed by the conventional BPSG film by making an annealing temperature into 850 degrees C or less, at low temperature, while flattening of a layer insulation film is possible, -izing of the 1st and 2nd silicon oxides can fully be carried out [precise]. Moreover, if an annealing temperature is performed at the temperature exceeding 850 degrees C, the source and a drain diffusion layer will be expanded more than required, problems, such as a punch through, will be started, and detailed-ization of an element will become difficult.

[0026] By forming the 2nd porous silicon oxide on the 1st silicon oxide Even if there is a rapid temperature change in annealing processing at a process (c) like [at the time of putting a wafer directly on the bottom of the temperature of 600-850 degrees C] Annealing processing can be performed without producing a crack in this 1st silicon oxide, since the 2nd silicon oxide of the above has moderate softness and can absorb the stress of the 1st silicon oxide.

[0027] In order to prevent more certainly that a crack arises in the 1st silicon oxide, as for the annealing processing in the aforementioned process (c), it is desirable to be carried out by run ping annealing which raises temperature continuously or intermittently.

[0028] In this invention, it is desirable to form the silicon oxide which at least one sort of the compound containing a silicon compound, oxygen, and oxygen is made to react by the chemical-vapor-deposition method, and serves as a base layer before the aforementioned process (a). This base layer has the passivation function which neither moisture nor an excessive impurity moves, and the function which raises the adhesion of a silicon substrate and the 1st silicon oxide from the 1st silicon oxide in the silicon substrate which is the lower layer.

[0029] Moreover, in the manufacture method concerning this invention, the through hole of the shape of a taper to which aperture becomes small gradually toward a pars basilaris ossis occipitalis is obtained from the upper-limit section in the layer insulation film obtained by the manufacture method mentioned above. That is, the through hole which the 1st silicon oxide of the above does not have a level difference since, as for the 1st silicon oxide and 2nd silicon oxide, the etch rate has touched good by both interface slightly small compared with the 2nd silicon oxide, and has a moderate straight-line-like taper is formed. In the through hole of the shape of such a taper, an aluminum film or an aluminium alloy film can be embedded by the spatter, and the contact structure excellent in conductivity can be formed, for example.

[0030] Although the aforementioned through hole was formed of the dry etching of an anisotropy, it may make the upper-limit section of a through hole form in others in the shape of [which curved further] a taper combining isotropic wet etching and the dry etching of an anisotropy.

[0031] Moreover, it is desirable to form the 1st aluminum film which consists of an alloy which makes aluminum or aluminum a principal component at the temperature of 200 degrees C or less first in the aforementioned through hole, and to form after that the 2nd aluminum film which consists of an alloy which makes aluminum or aluminum a principal component at the temperature of 300 degrees C or more.

[0032] As an alloy which makes the aforementioned aluminum a principal component, at least one sort chosen from copper, silicon, germanium, magnesium, cobalt, beryllium, etc. of 2 yuan or 3 yuan or more alloys can be illustrated.

[0033] The semiconductor device formed by the above manufacture method The layer insulation film formed on the semiconductor substrate containing an element, and the aforementioned semiconductor substrate, The barrier layer formed in the front face of the through hole formed in the aforementioned layer insulation film, the aforementioned layer insulation film, and the aforementioned through hole, And including the electric conduction film formed on the aforementioned barrier layer, the aforementioned layer insulation film is formed on the 1st silicon oxide formed of the polycondensation reaction of a silicon compound and a hydrogen peroxide, and the 1st silicon oxide of the above, and contains the 2nd silicon oxide containing an impurity.

[0034]

[Embodiments of the Invention] Drawing 1 - drawing 4 are the outline cross sections for explaining the manufacture method of the semiconductor device concerning this invention, and the gestalt of 1 operation of a semiconductor device. Drawing 1 (A) The process for - (C) and drawing 2 (A), and (B) manufacturing the wiring field L1 of the 1st layer, and drawing 3 (A), (B) and drawing 4 (A), and (B) manufacturing the wiring field L2 of the 2nd layer is shown.

[0035] Below, an example of the manufacture method of a semiconductor device is shown.

[0036] (A) Explain the process shown in drawing 1 (A).

[0037] (Formation of an element) The MOS device is formed in a silicon substrate 11 by the method generally used first. Specifically, the field insulator layer 12 is formed of selective oxidation on a silicon substrate 11, and the gate oxide film 13 is formed in an active field. The gate electrode 14 is formed by carrying out the spatter of the tungsten silicide on the polysilicon contest film into which SiH₄ was pyrolyzed and was grown up, carrying out the laminating of the silicon oxide 18 further, and *****ing to a predetermined pattern further by channel pouring, after adjusting threshold voltage. At this time, the wiring layer 37 which consists of a polysilicon contest film and a tungsten silicide film is formed on the field insulator layer 12 if needed.

[0038] Subsequently, the low concentration impurity layer 15 of a source field or a drain field is formed by carrying out the ion implantation of Lynn. Subsequently, after the side-attachment-wall spacer 17 which becomes the side of the gate electrode 14 from a silicon oxide is formed, the high concentration impurity layer 16 of a source field or a drain field is formed by carrying out the ion implantation of the arsenic and activating an impurity by annealing processing using the halogen lamp.

[0039] Next, a predetermined silicon-substrate field is exposed by forming a vapor-growth silicon oxide 100nm or less, and *****ing this film alternatively with the mixed-water solution of HF and NH₄F. then -- for example, the silicon-substrate

front face which carried out opening by carrying out the spatter of the titanium by about 30-100nm thickness, and performing moment annealing for several seconds - about 60 seconds at the temperature of 650-750 degrees C into the nitrogen-gas-atmosphere mind which controlled oxygen to 50 ppm or less -- the monochrome silicide layer of titanium -- a silicon-oxide 18 top -- titanium -- a rich titanium nitride RAIDO (TiN) layer is formed Subsequently, if immersed into NH₄OH and the mixed-water solution of H₂O₂, etching removal of the aforementioned titanium nitride RAIDO layer will be carried out, and the monochrome silicide layer of titanium will remain only in a silicon-substrate front face. Furthermore, perform 750-850-degree C lamp annealing, the aforementioned monochrome silicide layer is made to form into die silicide, and the titanium silicide layer 19 is formed in the front face of the high concentration impurity layer 16 at a self-adjustment target. [0040] In addition, when the gate electrode 14 is formed only with contest polysilicon and it is made to expose by selective etching, both a gate electrode, source, and drain field become the CHITANSA LISA id structure separated with the side-attachment-wall spacer.

[0041] In addition, the Salicide structure may consist of tungsten silicide and molybdenum silicide instead of titanium silicide.

[0042] (B) Next, explain the process shown in drawing 1 (B).

[0043] (Formation of the 1st layer insulation film I1) The 1st layer insulation film I1 consists of the 4th silicon oxide 20, the 1st silicon oxide 22, the 2nd silicon oxide 24, and the 3rd silicon oxide 26 in order of the silicon oxide of four layers, i.e., a lower shell.

[0044] a. **** of the 4th silicon oxide 20 -- the 4th silicon oxide 20 of 100-200nm of thickness is first formed by making a tetrapod ethoxy run (TEOS) and oxygen react by the plasma-chemistry vapor-growth (CVD) method at 300-500 degrees C This silicon oxide 20 does not have oxidation or the dregs ping of the silicide layer 19, either, and highly [insulation], its etch rate to the solution of hydrogen fluoride is also slow, and it turns into a precise film from the film grown up from SiH₄.

[0045] Since a oxidizing gas and titanium silicide will tend to produce a crack and ablation in response to the early stages of membrane formation here if the membrane formation temperature at this time is high although the direct silicon oxide 20 is made to form on the titanium silicide layer 19, as for processing temperature, it is preferably desirable to carry out at 250-400 degrees C more preferably 600 degrees C or less. And if it is annealing and vapor-phase-oxidation processing which are exposed to oxidizing atmospheres other than a steam after [which the silicon oxide mentioned above in about 100nm thickness on the titanium silicide layer 19] being comparatively formed at low temperature, it will not become a problem even if it raises temperature to 900-degree-C grade.

[0046] b. Form the 2.5x10²Pa or less of the 1st silicon oxide 22 preferably formation of the 1st silicon oxide 22, next by making SiH₄ and H₂O₂ react by CVD by using nitrogen gas as a carrier more preferably under reduced pressure of 0.3x10² to 2.0x10²Pa. The 1st silicon oxide 22 is formed by the thickness which has larger thickness than the level difference of the 4th lower layer silicon oxide 20 at least, that is, fully covers this level difference. Moreover, the upper limit of the thickness of the 1st silicon oxide 22 is set as the grade which a crack does not produce in this film. Specifically, in order to obtain better flat nature, as for the thickness of the 1st silicon oxide 22, it is desirable that it is thicker than a lower layer level difference, and it is preferably set as 300-1000nm.

[0047] Since it participates in the fluidity at the time of membrane formation of this film, membrane fluidity will fall if membrane formation temperature is high, and the membrane formation temperature of the 1st silicon oxide 22 spoils flat nature, 0-20 degrees C of temperature at the time of membrane formation are more preferably set as 0-10 degrees C.

[0048] Moreover, although especially the flow rate of H₂O₂ is not restricted, it is desirable that concentration is 55 to 65 volume %, for example, and it is a flow rate more than the double precision of SiH₄, and it is desirable to be set, for example as the flow rate range of 100 - 1000SCCM by gas conversion from membranous homogeneity and the point of a throughput.

[0049] The 1st silicon oxide 22 formed at this process is in the state of silanol polymer, and a fluidity is good and has a high self-flattening property. Moreover, since many hydroxyl groups (-OH) are included, the 1st silicon oxide 22 has hygroscopicity in a high state.

[0050] c. Put under existence of SiH₄, PH₃, and N₂O continuously after leaving it for 30 - 120 seconds under reduced pressure within formation, next the chamber of the 2nd silicon oxide 24 and removing some moisture in the 1st silicon oxide 22. By making gas react by the plasma CVD method at the temperature of 300-450 degrees C at 200-600kHz high frequency, the PSG film (the 2nd silicon oxide) 24 of 100-600nm of thickness is formed. As for this 2nd silicon oxide 24, it is desirable to be formed after being saved in the atmosphere in which the hygroscopicity of the 1st silicon oxide 22 of the above is continuously formed in in consideration of a high thing following formation of the 1st silicon oxide 22 of the above, or the 1st silicon oxide 22 does not contain moisture.

[0051] Moreover, the 2nd silicon oxide 24 needs that desorption of gasification components, such as water contained in the 1st silicon oxide 22 of the above by the annealing processing performed behind and hydrogen, is easy, and to be porous (porosity) in consideration of fully being carried out. For that purpose, temperature is desirable and it is preferably [more] desirable [the 2nd silicon oxide 24] 1MHz or less preferably to form membranes by the 200-600kHz plasma CVD method more preferably, and to include impurities, such as Lynn, 300-400 degrees C 450 degrees C or less. By containing such an impurity in the 2nd silicon oxide 24, it will be in a more nearly porous state and the 2nd silicon oxide 24 not only can ease the stress to a film, but can have the gettering effect over alkali ion etc. with it. The concentration of such an impurity is set up in consideration of points, such as the gettering effect and stress-proof nature. For example, when an impurity is Lynn, it is desirable to be contained at 2 - 6% of the weight of a rate.

[0052] Moreover, in plasma CVD, desorption of the hydrogen bond in the 1st silicon oxide 22 is promoted by using N₂O as a compound containing oxygen. Consequently, gasification components, such as moisture contained in the 1st silicon oxide 22 and

hydrogen, can be removed more certainly.

[0053] In consideration of the role which adjusts the thickness of the layer insulation film needed, and the function by which N₂O plasma is desorbed from hydrogen bond, 100nm or more of thickness of this 2nd silicon oxide 24 is more preferably set as 100-600nm.

[0054] d. Perform annealing processing at the temperature of 600-850 degrees C in annealing processing, next nitrogen-gas-atmosphere mind. By this annealing processing, the 1st silicon oxide 22 of the above and the 2nd silicon oxide 24 turn precisely, and have good insulation and good water resistance. That is, by setting an annealing temperature as 600 degrees C or more, the condensation polymerization reaction of the silanol in the 1st silicon oxide 22 is performed nearly completely, and the water and hydrogen which are contained in this film are fully emitted, and can form a precise film. Moreover, by setting an annealing temperature as 850 degrees C or less, it cannot have bad influences, such as a punch through and junction leak, on the diffusion layer of the source field which constitutes an MOS transistor, or a drain field, and detailed-ization of an element can be attained.

[0055] In annealing processing, in order to make small influence of a heat strain to the 1st silicon oxide 22, it is desirable to perform run ping annealing which raises the temperature of a wafer gradually or continuously. For example, when carrying out a temperature up to an annealing temperature (600-850 degrees C) after keeping a wafer warm at about 400 degrees C, high impurity concentration of the 2nd silicon oxide 24 can be made quite low. For example, when an impurity is Lynn, the gettering effect of mobile ion is set aside and the concentration of Lynn is checking that a crack does not arise in the 1st silicon oxide 22 at least 2 or less % of the weight.

[0056] e. Form the 3rd silicon oxide 26 of 1000-1500nm of thickness by the plasma CVD method at 350-400 degrees C using formation, next the 3rd TEOS and oxygen of a silicon oxide 26.

[0057] Even when not performing annealing, the silicon oxide of the TEOS-oxygen using the plasma CVD method is of the same grade as the 1st silicon oxide 22 of the above which carried out elevated-temperature annealing, and the 2nd silicon oxide 24, or has a somewhat quick dry etching speed. This becomes the factor which obtains the contact hole of a good configuration, without producing the vena contracta and a level difference on the hole side in formation of the contact hole mentioned later.

[0058] (C) Next, explain the process shown in drawing 1 (C).

[0059] (Smoothing by CMP) Subsequently, the 3rd silicon oxide 26 of the above, and if needed, by the CMP method, predetermined thickness is ground and the 2nd silicon oxide 24 of the above and the 1st silicon oxide 22 are smoothed. And the 1st silicon oxide 22 of the above, the 2nd silicon oxide 24, and the 3rd silicon oxide 26 can obtain a flat front face, though a part of 2nd silicon oxide 24 or 1st silicon oxide 22 is exposed to a front face with polish, since polish speed is almost the same, therefore management of the amount of polishes is easy for them.

[0060] For example, according to research of this invention persons, the polish speed of each silicon oxide was as follows.

[0061]

The 1st silicon oxide (800 degrees C of annealing temperatures) A part for;250nm/ The 2nd silicon oxide (800 degrees C of annealing temperatures) A part for;250nm/ The 3rd silicon oxide (with no annealing) A part for;250nm/ The BPSG film for comparison (900 degrees C of annealing temperatures); the process shown in 350nm a part (D) for /next and drawing 2 (A) is explained.

[0062] (Formation of a contact hole) The contact hole 32 whose aperture is 0.2-0.5 micrometers is formed by carrying out anisotropic etching of the silicon oxides 20, 22, 24, and 26 which subsequently constitute the 1st layer insulation film 11 from a reactant ion etcher which made CHF₃ and CF₄ the main gas alternatively.

[0063] This contact hole 32 constitutes the shape of a taper to which aperture becomes small linearly toward a pars basilaris ossis occipitalis from the upper-limit section. Although the angle theta of a taper cannot generally be ***** (ed) according to etching conditions etc., it has five - 15 inclinations, for example. As a reason the through hole of the shape of such a taper is obtained, silicon oxides 20, 22, 24, and 26 have the fundamental almost same etch rate, and the interface of each silicon oxide is [1st] further that the 1st silicon oxide 22 has a slightly small etch rate compared with the 2nd silicon oxide 24, and the 2nd to have stuck very good. Within the contact hole 32 of the shape of such a taper, good deposition of an aluminum film is possible so that it may mention later.

[0064] The dry etching speed of each silicon oxide which invention-in-this-application persons measured below is indicated. In addition, dry etching was performed on condition that power;800W, atmospheric pressure;20Pa, and etchant gas;CF₄:CHF₃:helium=1:2:9.

[0065]

The 1st silicon oxide (800 degrees C of annealing temperatures) A part for;525nm/ The 2nd silicon oxide (800 degrees C of annealing temperatures) A part for;550nm/ The 3rd silicon oxide (with no annealing) A part for;565nm/ The BPSG film for comparison (900 degrees C of annealing temperatures); the process shown in 750nm a part (E) for /next and drawing 2 (B) is explained.

[0066] (Degasifying processing) ***** which includes a degasifying process first -- it ***** just

[0067] a lamp chamber -- the base pressure of 1.5×10^{-4} or less Pa -- 150-350 degrees C (heat treatment A) of lamp heating for 30 - 60 seconds are preferably given at the temperature of 150-250 degrees C Subsequently, degasifying processing is performed by introducing argon gas by another chamber by the pressure of 1×10^{-1} to 15×10^{-1} Pa, and performing heat treatment for 30 - 300 seconds (degasifying process; heat treatment B) at the temperature of 150-550 degrees C.

[0068] In this process, the moisture adhering to the wafer etc. is removable by mainly heat-treating the whole wafer including the rear face and the side of a wafer in heat treatment A first.

[0069] Furthermore, in heat treatment B, the gasification component in the 1st silicon oxide 22 which constitutes the 1st layer insulation film 11 (H, H₂O) is mainly removable. Consequently, generating of the gasification component from the 1st layer insulation film 11 can be prevented at the time of formation of the barrier layer of the following process, and an aluminum film.

[0070] The barrier layer 33 is constituted in the gestalt of this operation by the multilayer which consists of a barrier film which has barrier ability, and an electric conduction film. An electric conduction film is formed between a barrier film and an impurity diffusion layer, in order to raise conductivity with the impurity diffusion layer formed in the barrier film and the silicon substrate, i.e., a source field, and a drain field. As a barrier film, the common matter, for example, titanium nitride (TiN) and a titanium tungsten, can be used preferably. Moreover, refractory metals, such as titanium, cobalt, and a tungsten, can be used as an electric conduction film. These titanium, cobalt, and a tungsten react with the silicon which constitutes a substrate, and serve as silicide.

[0071] Since it dissolves the gasification component (O, H, H₂O, N) of dozens atom %, before a barrier layer, for example, a TiN film / Ti film, forms these films, it is very effective [film] to remove the gasification component in the 1st layer insulation film 11, when forming the aluminum film within a contact hole good. If the gasification component in the 1st [of the low rank of a barrier layer] layer insulation film 11 is not fully removed, at the temperature at the time of formation of a barrier layer (usually 300 degrees C or more), the gasification component in the 1st layer insulation film 11 will be emitted, and this gas will be incorporated in a barrier layer. Furthermore, in order that this gas may secede from a barrier layer at the time of membrane formation of an aluminum film and may come out to the interface of a barrier layer and an aluminum film, it has a bad influence on the adhesion of an aluminum film, or a fluidity.

[0072] (Membrane formation of a barrier layer) By the sputter, as an electric conduction film which constitutes the barrier layer 33, a titanium film is formed by 20-70nm thickness, and, subsequently a TiN film is formed as a barrier film by another chamber at 30-150nm thickness. The temperature of a sputter is chosen in 200-450 degrees C according to thickness.

[0073] Next, titanium oxide can be formed in the shape of an island into a barrier layer by exposing for 10 - 100 seconds into oxygen plasma by the pressure of 0.1x10² to 1.5x10²Pa, and carrying out annealing processing over 10 - 60 minutes in 450-700-degree C nitrogen or hydrogen atmosphere. It is checking that the barrier property of a barrier layer can be raised by this processing.

[0074] Moreover, 400-800-degree C heat treatment in the lamp annealing furnace which contains hundreds of ppm - several% of oxygen at least can also perform this annealing processing, and the barrier property of a barrier layer can be raised similarly.

[0075] In addition, although illustration is not carried out, it is the purpose which raises the wettability to the aluminum film mentioned later, and the WETTENGU layer which consists of titanium, cobalt, silicon, etc. may be formed in the front face of the barrier layer 33. The 1st aluminum membrane fluidity can be raised by preparing such a WETTENGU layer. There should just usually be thickness dozens of nm or more of a WETTENGU layer.

[0076] (Degasifying processing before membrane formation of an aluminum film, and cooling of a wafer) First, before cooling a wafer, heat treatment for 30 - 60 seconds (heat treatment C) is performed in a lamp chamber at the base pressure of 1.5x10⁻⁴ or less Pa, and the temperature of 150-250 degrees C, and matter, such as water adhering to the substrate, is removed. Then, before forming an aluminum film, 100 degrees C or less of substrate temperature are preferably lowered to ordinary temperature -50 degree C temperature. This cooling process is important in order to lower the substrate temperature which rose with the above-mentioned heat treatment C, for example, on the stage which has a water-cooled function, lays a wafer and lowers this wafer temperature to predetermined temperature.

[0077] Thus, in case the 1st aluminum film is formed by cooling a wafer, the 1st layer insulation film 11 and the barrier layer 33, and capacity further emitted from the whole wafer surface can be lessened as much as possible. Consequently, the influence of gas detrimental to the coverage nature and adhesion which stick to the interface of the barrier layer 33 and the 1st aluminum film 34 can be prevented.

[0078] (Membrane formation of an aluminum film) First, it is 30-100 degrees C in temperature more preferably, and the aluminum containing 0.2 - 1.0% of the weight of copper is formed at high speed by the sputter by 150-300nm of thickness, and 200 degrees C or less of 1st aluminum film 34 are formed. Then, it heats in substrate temperature of 420-460 degrees C within the same chamber, the aluminum which contains copper similarly is formed by the low speed by the sputter, and the 2nd aluminum film 35 of 300-600nm of thickness is formed. Here, in membrane formation of an aluminum film, although neither membrane formation conditions nor the design matter of a device manufactured can prescribe "high speed" generally, a sputtering rate 10nm [/second] or more is meant about, and a "low speed" means a sputtering rate 3nm [/second] or less about.

[0079] An example of the sputtering system for forming the 1st and 2nd aluminum films 34 and 35 to drawing 5 is shown. This sputtering system has the electrode 52 which serves both as the target 51 which serves as an electrode in a chamber 50, and a stage, and it is constituted so that the substrate (wafer) W processed may be installed on an electrode 52. The 1st gas supply way 53 is connected to a chamber 50, and the 2nd gas supply way 54 is connected to the electrode 52. Argon gas is supplied from each] the gas supply ways 53 and 54. And the temperature of Wafer W is controlled by the gas supplied from the 2nd gas supply way 54. In addition, the means for discharging the gas in a chamber 50 is not illustrated.

[0080] An example which controlled substrate temperature using such a sputtering system is shown in drawing 6. In drawing 6, a horizontal axis shows elapsed time and a vertical axis shows substrate (wafer) temperature. Moreover, in drawing 6, the line which the line shown with Sign a shows the substrate temperature change when setting the temperature of the stage 52 of a sputtering system as 350 degrees C, and is shown with Sign b shows change of the substrate temperature when raising the temperature of a stage 52 by supplying hot argon gas in a chamber through the 2nd gas supply way 54.

[0081] For example, the temperature control of a substrate is performed as follows. First, the temperature of a stage 52 is beforehand set as the temperature (350-500 degrees C) for forming the 2nd aluminum film. In case the 1st aluminum film is

formed, there is no supply of the gas from the 2nd gas supply way 54, and substrate temperature rises gradually by heating by the stage 52, as the sign a of drawing 6 shows. By supplying the gas heated through the 2nd gas supply way 54, in case the 2nd aluminum film is formed, substrate temperature rises rapidly and is controlled to become fixed at predetermined temperature so that the sign b of drawing 6 shows.

[0082] In the example shown in drawing 6, stage temperature is set as 350 degrees C, while substrate temperature is set as 125-150 degrees C, the 1st aluminum film 34 is formed, and membrane formation of the 2nd aluminum film 35 is performed immediately after that.

[0083] In membrane formation of an aluminum film, control of the power impressed to a sputtering system with membrane formation speed and a substrate temperature control is also important. That is, although membrane formation speed is related, in case membrane formation of the 1st aluminum film 34 is performed by high power, the 2nd aluminum film 35 is performed by low power and it switches to low power from still higher power, it is important not to make power into zero. If power is made into zero, an oxide film will be formed in the bottom of reduced pressure on the front face of the 1st aluminum film, the wettability of the 2nd [to the 1st aluminum film] aluminum film will fall, and both adhesion will become bad. In other words, by always impressing power, supplying activity aluminum to the front face of the aluminum film under membrane formation can be continued, and formation of an oxide film can be suppressed. In addition, although the size of power cannot generally be specified depending on a sputtering system, membrane formation conditions, etc., in the case of the temperature conditions shown, for example in drawing 6, it is desirable [a size] to set 5-10kW and low power as 300W-1kW for high power.

[0084] Thus, by forming continuously the 1st aluminum film 34 and the 2nd aluminum film 35 within the same chamber, control of temperature and power can be performed strictly and it becomes possible to form efficiently the aluminum film which is low temperature and was stabilized rather than before.

[0085] The thickness of the aluminum film 34 of the above 1st has desirable 200-400nm, for example, although the proper range is chosen from that a continuation layer can be formed and this aluminum film 34 by good step coverage in consideration of the ability to suppress discharge of the gasification component from the lower layer barrier layer 33 and the 1st layer insulation film 11. Moreover, although the 2nd aluminum film 35 is determined by the size of a contact hole, its aspect ratio, etc., in order for an aspect ratio to fill a hole 0.5 micrometers or less about by three, for example, 300-1000nm thickness is required for it.

[0086] (Membrane formation of an antireflection film) The antireflection film 36 of 30-80nm of thickness is formed by depositing TiN by the sputter by still more nearly another sputter chamber. Then, the deposit which consists of the aforementioned barrier layer 33, the 1st aluminum film 34, the 2nd aluminum film 35, and an antireflection film 36 by the anisotropy dry etcher which makes the gas of Cl₂ and BCl₃ a subject is *****ed alternatively, and patterning of the 1st metal wiring layer 30 is performed.

[0087] Thus, in the formed metal wiring layer 30, it was checked that aluminum is embedded by good step coverage, without an aspect ratio generating a void in 0.5-3 in the contact hole whose aperture is 0.2-0.8 micrometers.

[0088] (F) Next, explain the process shown in drawing 3 (A).

[0089] (Formation of the 2nd layer insulation film 12) The 2nd layer insulation film 12 has the same composition as the layer insulation film 11 of the above 1st fundamentally. That is, the 2nd layer insulation film 12 consists of the 8th silicon oxide 70, the 5th silicon oxide 72, the 6th silicon oxide 74, and the 7th silicon oxide 76 in order of the silicon oxide of four layers, i.e., a lower shell. And these silicon oxides 70, 72, 74, and 76 are formed by the same method as the aforementioned silicon oxides 20, 22, 24, and 26 except annealing processing. Although main portions are explained below, a publication is omitted about a common matter.

[0090] a. **** of the silicon oxide 70 of the octavus -- the silicon oxide 70 of the octavus of 50-200nm of thickness is first formed by making a tetrapod ethoxy run (TEOS) and oxygen react by the plasma-chemistry vapor-growth (CVD) method at 300-500 degrees C

[0091] b. Form the 2.5x10²Pa or less of the 5th silicon oxide 72 preferably formation of the 5th silicon oxide 72, next by making SiH₄ and H₂O₂ react by CVD at the temperature of 0-10 degrees C by using nitrogen gas as a carrier more preferably under reduced pressure of 0.3x10² to 2x10²Pa. The 5th silicon oxide 72 is formed by the thickness which has larger thickness than the level difference of the 8th lower layer silicon oxide 70 at least, that is, fully covers this level difference like the 1st silicon oxide 22 of the above. Moreover, the upper limit of the thickness of the 5th silicon oxide 72 is set as the grade which a crack does not produce in this film. Specifically, in order to obtain better flat nature, as for the thickness of the 5th silicon oxide 72, it is desirable that it is thicker than a lower layer level difference, and it is preferably set as 500-1000nm.

[0092] 0-20 degrees C of membrane formation temperature of the 5th silicon oxide 72 are more preferably set as 0-10 degrees C.

[0093] The 5th silicon oxide 72 formed at this process has a high fluidity, and is excellent in a flattening property.

[0094] c. The PSG film (the 6th silicon oxide) 74 of 100-600nm of thickness is formed by making it react to the bottom of existence of formation of the 6th silicon oxide 74 next SiH₄ and PH₃, and N₂O by the plasma CVD method at the temperature of 300-450 degrees C at 200-600kHz high frequency.

[0095] Moreover, the 6th silicon oxide 74 needs that desorption of gasification components, such as water contained in the 5th silicon oxide 72 of the above like the 2nd silicon oxide 24 of the above by the annealing processing performed behind, is easy, and to be porous (porosity) in consideration of fully being carried out. For that purpose, temperature is desirable and it is preferably [more] desirable [the 6th silicon oxide 74] 1MHz or less preferably to form membranes by the 200-600kHz RF plasma CVD method more preferably, and to contain impurities, such as Lynn, 300-400 degrees C 450 degrees C or less. By containing such an impurity in the 2nd silicon oxide 74, the 2nd silicon oxide 74 will be in a more nearly porous state, and can ease the stress to a film. The concentration of such an impurity is set up in consideration of points, such as stress-proof nature and

the gettering effect. For example, when an impurity is Lynn, it is desirable to be contained at 1 - 6% of the weight of a rate.

[0096] Moreover, in plasma CVD, desorption of the hydrogen bond in the 5th silicon oxide 72 is promoted by using N₂O as a compound containing oxygen. Consequently, gasification components, such as moisture contained in the 5th silicon oxide 72, can be removed more certainly.

[0097] 100nm or more of thickness of this 6th silicon oxide 74 is more preferably set as 200-600nm.

[0098] d. Perform annealing processing at annealing processing, next the temperature of 350-450 degrees C. By this annealing processing, the 5th silicon oxide 72 of the above and the 6th silicon oxide 74 turn precisely, and have good insulation and good water resistance. That is, by setting an annealing temperature as 350 degrees C or more, the condensation polymerization reaction of the silanol in the 5th silicon oxide 72 is performed nearly completely, and the moisture contained in this film is fully emitted, and can form a precise film. Moreover, it does not have a bad influence on the aluminum film which constitutes the 1st wiring layer 30 by setting an annealing temperature as 450 degrees C or less.

[0099] e. Form the 3rd silicon oxide 76 of 1000-1500nm of thickness by the plasma CVD method at 350-400 degrees C using formation, next the 7th TEOS and oxygen of a silicon oxide 76.

[0100] (G) Next, explain the process shown in drawing 3 (B).

[0101] (Smoothing by CMP) The 6th silicon oxide 74 of the above and the 5th silicon oxide 72 are ground and smoothed by predetermined thickness by the CMP method the 7th silicon oxide 76 of the above, and if needed. By this data smoothing, though a part of 7th silicon oxide 74 or 5th silicon oxide 72 is exposed to a front face with polish, a flat front face can be obtained, therefore management of the amount of polishes is easy.

[0102] (H) Next, explain the process shown in drawing 4 (A).

[0103] (Formation of a beer hall) By carrying out anisotropic etching of the 2nd layer insulation film I2 and antireflection film 36 alternatively by the reactant ion etcher which made CHF₃ and CF₄ the main gas, the beer hall 62 whose aperture is 0.3-0.5 micrometers is formed.

[0104] This beer hall 62 constitutes the shape of a taper to which aperture becomes small gradually toward a bottom from the upper-limit section like the aforementioned contact hole 32. Although the angle theta of a taper cannot generally be ***** (ed) according to etching conditions etc., it has five - 15 inclinations, for example.

[0105] (I) Next, the process shown in drawing 4 (B) is explained.

[0106] (Degasifying processing) ***** which includes a degasifying process first -- it ***** just

[0107] a lamp chamber -- the base pressure of 1.5×10^{-4} or less Pa -- 150-350 degrees C (heat treatment D) of lamp heating for 30 - 60 seconds are preferably given at the temperature of 150-250 degrees C. Subsequently, degasifying processing is performed by introducing argon gas by another chamber by the pressure of 1×10^{-1} to 15×10^{-1} Pa, and performing heat treatment for 30 - 300 seconds (degasifying process; heat treatment E) at the temperature of 300-500 degrees C.

[0108] In this process, the moisture adhering to the wafer etc. is removable by mainly heat-treating the whole wafer including the rear face and the side of a wafer in heat treatment D first.

[0109] Furthermore, in heat treatment E, the gasification component in the 2nd layer insulation film I2 (H, H₂O) is mainly removable. Consequently, generating of the gasification component from the 2nd layer insulation film I2 can be prevented at the time of formation of the WETTENGU layer of the following process, and an aluminum film.

[0110] In the form of this operation, since it dissolves the gasification component (O, H, H₂O, N) of dozens atom %, before a WETTENGU layer, for example, Ti film, forms this film, it is very effective [film] to remove the gasification component in the 2nd layer insulation film I2, when forming the aluminum film in a beer hall good. If the gasification component in the 2nd [of the low rank of a WETTENGU layer] layer insulation film I2 is not fully removed, at the temperature at the time of formation of a WETTENGU layer (usually 300 degrees C or more), the gasification component in the 2nd layer insulation film I2 will be emitted, and this gas will be incorporated in a WETTENGU layer. Furthermore, in order that this gas may secede from a WETTENGU layer at the time of membrane formation of an aluminum film and may come out to the interface of a WETTENGU layer and an aluminum film, it has a bad influence on the adhesion of an aluminum film, or a fluidity.

[0111] (Membrane formation of a WETTENGU layer) By the spatter, a titanium film is formed by 20-70nm thickness as a film which constitutes the WETTENGU layer 63. The temperature of a spatter is chosen in 200-450 degrees C according to thickness.

[0112] (Degasifying processing before membrane formation of an aluminum film, and cooling of a wafer) First, before cooling a wafer, heat treatment for 30 - 60 seconds (heat treatment F) is performed in a lamp chamber at the base pressure of 1.5×10^{-4} or less Pa, and the temperature of 150-250 degrees C, and matter, such as water adhering to the substrate, is removed. Then, before forming an aluminum film, 100 degrees C or less of substrate temperature are preferably lowered to ordinary temperature -50 degree C temperature. This cooling process is important in order to lower the substrate temperature which rose with the above-mentioned heat treatment F, for example, on the stage which has a water-cooled function, lays a wafer and lowers this wafer temperature to predetermined temperature.

[0113] Thus, in case the 1st aluminum film is formed by cooling a wafer, the 2nd layer insulation film I2 and the WETTENGU layer 63, and capacity further emitted from the whole wafer surface can be lessened as much as possible. Consequently, the influence of gas detrimental to the coverage nature and adhesion which stick to the interface of the WETTENGU layer 63 and the 1st aluminum film 64 can be prevented.

[0114] (Membrane formation of an aluminum film) First, it is 30-100 degrees C in temperature more preferably, and the aluminum containing 0.2 - 1.0% of the weight of copper is formed at high speed by the spatter by 150-300nm of thickness, and 200 degrees C or less of 1st aluminum film 64 are formed. Then, it heats in substrate temperature of 420-460 degrees C within

the same chamber, the aluminum which contains copper similarly is formed by the low speed by the spatter, and the 2nd aluminum film 65 of 300-600nm of thickness is formed.

[0115] The thing same as a sputtering system as the equipment shown in drawing 5 can be used. About the composition of the aforementioned sputtering system, the temperature control of a wafer, and the power at the time of a spatter, since it is the same as that of the case of the 1st metal wiring layer 30, detailed explanation is omitted.

[0116] By forming continuously the 1st aluminum film 64 and the 2nd aluminum film 65 within the same chamber, control of temperature and power can be performed strictly and it becomes possible to form efficiently the aluminum film which is low temperature and was stabilized rather than before.

[0117] The thickness of the aluminum film 64 of the above 1st has desirable 100-300nm, for example, although the proper range is chosen from that a continuation layer can be formed and this aluminum film 64 by good step coverage in consideration of the ability to suppress discharge of the gasification component from the lower layer WETTENGU layer 63 and the 2nd layer insulation film I2. Moreover, although the 2nd aluminum film 65 is determined by the size of a beer hall 62, its aspect ratio, etc., in order for an aspect ratio to fill a hole 0.5 micrometers or less about by three, for example, 300-800nm thickness is required for it.

[0118] (Membrane formation of an antireflection film) The antireflection film 66 of 30-80nm of thickness is formed by depositing TiN by the spatter by still more nearly another spatter chamber. Then, the deposit which consists of the aforementioned WETTENGU layer 63, the 1st aluminum film 64, the 2nd aluminum film 65, and an antireflection film 66 by the anisotropy dry etcher which makes the gas of Cl₂ and BCl₃ a subject is *****ed alternatively, and patterning of the 2nd metal wiring layer 60 is performed.

[0119] Thus, in the formed metal wiring layer 60, it was checked that aluminum is embedded by good step coverage, without an aspect ratio generating a void in 0.5-3 in the beer hall whose aperture is 0.2-0.8 micrometers.

[0120] Henceforth, it is the 3rd and the 4th like the 2nd wiring field L2 if needed. -- A multilayer-interconnection field can be formed.

[0121] In the form of this operation, the following things can be considered as a reason for having the flat nature excellent in the 1st and 2nd layer insulation films I1 and I2.

[0122] (a) Since the resultant containing a silanol formed of the reaction of a silicon compound and a hydrogen peroxide has a high fluidity, when the irregularity on the front face of a wafer forms these films, flattening of the 1st silicon oxide 22 formed at the process shown in drawing 1 (B) and drawing 3 (A) and the 5th silicon oxide 72 is carried out highly.

[0123] (b) Each silicon oxide which constitutes the 1st and 2nd layer insulation films I1 and I2, since the 5th, the 6th, and 7th silicon oxides 72, 74, and 76 have a polish speed of the same grade in CMP in the 1st, the 2nd, and 3rd silicon oxides 22 and 24 and 26 rows especially, even if it is the case where a silicon oxide which is different on a front face lives together partially, good flat nature is obtained.

[0124] Moreover, in the form of this operation, the following things can be considered as a reason the 1st and 2nd aluminum films 64 and 65 were embedded by the 1st and the 2nd aluminum film 34, and 35 rows good in the contact hole 32 and the beer hall 62, respectively.

[0125] (a) By gasifying water and nitrogen which are contained in the insulator layers I1 and I2 between each class by performing a degasifying process, and fully emitting In the 1st subsequent aluminum film 34 and 64 and membrane formation of the 2nd aluminum 35 and 65, by preventing generating of the gas from the layer insulation films I1 and I2, the barrier layer 33, or the WETTENGU layer 63 Raised the adhesion of the WETTENGU layer 63 and the 1st aluminum film 64 to the barrier layer 33, the 1st aluminum film 34, and the row, and membrane formation of good step coverage was possible for.

[0126] (b) what the adhesion of the 1st aluminum film 34 and 64 was raised for in addition to the effect of the aforementioned degasifying process as the moisture or nitrogen which are contained in the WETTENGU layer 63 in substrate temperature at the layer insulation films I1 and I2 and barrier layer 33 row 200 degrees C or less by setting it as low temperature comparatively were not made to emit in membrane formation of the 1st aluminum film 34 and 64

[0127] (c) Since the 1st aluminum film 34 and 64 the very thing play further the role which suppresses generating of the gas from a lower layer when substrate temperature goes up, the 2nd following aluminum film 35 and 65 can be formed at comparatively high temperature, and flow diffusion of the 2nd aluminum film can be performed good.

[0128] By the above method, the semiconductor device (refer to drawing 4 (B)) concerning this invention can be formed. This semiconductor device has the 1st wiring field L1 formed on the silicon substrate 11 which contains the MOS device at least, and the aforementioned silicon substrate 11.

[0129] The wiring field L1 of the above 1st is formed on the 4th silicon oxide 20 used as a base layer, the 1st silicon oxide 22 formed of the polycondensation reaction of a silicon compound and a hydrogen peroxide, and the 1st silicon oxide 22 of the above. It is formed on the 2nd silicon oxide 24 containing impurities, such as Lynn, and the 2nd silicon oxide 24 of the above. By CMP Were formed on the barrier layer 33 formed in the front face of the contact hole 32 formed in the 1st layer insulation film I1 which consists of the 3rd silicon oxide 26 by which flattening was carried out, and the aforementioned layer insulation film I1, the aforementioned layer insulation film I1, and the aforementioned contact hole 32, and the aforementioned barrier layer 33. It has the aluminum films 34 and 35 which consist of an alloy which makes aluminum or aluminum a principal component. And the aforementioned aluminum film 34 is connected to the titanium silicide layer 19 through the barrier layer 33.

[0130] The 2nd wiring field L2 formed on the wiring field L1 of the above 1st It is formed on the silicon oxide 70 of the octavus used as a base layer, the 5th silicon oxide 72 formed of the polycondensation reaction of a silicon compound and a hydrogen peroxide, and the 5th silicon oxide 72 of the above. It is formed on the 6th silicon oxide 74 containing impurities, such as Lynn,

and the 6th silicon oxide 74 of the above. By CMP Were formed on the WETTENGU layer 63 formed in the front face of the beer hall 62 formed in the 2nd layer insulation film I2 which consists of the 7th silicon oxide 76 by which flattening was carried out, and the aforementioned layer insulation film I2, the aforementioned layer insulation film I2, and the aforementioned beer hall 62, and the aforementioned WETTENGU layer 63. It has the aluminum films 64 and 65 which consist of an alloy which makes aluminum or aluminum a principal component.

[0131] As mentioned above, according to the gestalt of this operation, the layer insulation film which has very good flat nature can be formed by forming the silicon oxide containing a silanol obtained according to the gaseous phase reaction of a silicon compound and a hydrogen peroxide, and forming the silicon oxide in which flattening was further carried out to the best layer by CMP. Since especially the 1st layer insulation film can be formed at remarkable low temperature compared with the conventional BPSG film, it can improve a property in respect of a punch through, junction leak, etc., therefore can attain detailed-izing of an element, and reliable contact structure, and its manufacture process top is also advantageous. Moreover, since it has flat nature with an advanced layer insulation film, a process margin including processing of a wiring layer etc. can be made to be able to increase, and quality and the yield can be raised.

[0132] Furthermore, in the gestalt of this operation, by forming an aluminum film continuously [it is still more desirable and] within the same chamber including a degasifying process and a cooling process at least before the spatter of an aluminum film, it became possible to embed the contact hole and beer hall to about 0.2 micrometers only by aluminum or the aluminium alloy, and improvement was able to be aimed at in respect of reliability and the yield. Moreover, there are also no copper segregation and unusual growth of crystal grain in the aluminum film which constitutes the contact section, and the good thing was checked also in respect of reliability including migration etc.

[0133] (others -- gestalt of operation) this invention is not limited to the gestalt of the above-mentioned implementation, but can replace the part with the following meanses

[0134] (a) In the gestalt of the aforementioned implementation, although the dinitrogen oxide was used as a compound containing oxygen at the time of membrane formation by the plasma CVD of the 2nd silicon oxide 24 instead, ozone can also be used. And before forming the 2nd silicon oxide 24, it is desirable to expose a wafer to ozone atmosphere.

[0135] For example, Wafer W is laid and it is made to move at the rate of predetermined using the belt furnace shown in drawing 7 onto the conveyance belt 80 heated by 400-500 degrees C at the heater 82. At this time, ozone is supplied from 1st gas head 86a, and the aforementioned wafer W is passed for the inside of 2 - 8% of the weight of ozone atmosphere over the time for 5 minutes or more. Subsequently, ozone, TEOS, and TMP (P(OCH₃)₃) are mostly supplied by the ordinary pressure from the 2nd and 3rd gas heads 86b and 86c, and the PSG film (the 2nd silicon oxide) 24 whose concentration of Lynn is 3 - 6 % of the weight is formed by 100-600nm of thickness. In addition, in drawing 7, a sign 84 shows covering.

[0136] Thus, by using ozone instead of a dinitrogen oxide, the silicon oxide by TEOS can be formed by the ordinary pressure CVD. Moreover, membranes can be continuously formed efficiently by using a belt furnace.

[0137] Moreover, it was checked by the thermal-desorption spectrum (TDS) and the infrared spectroscopy (FTIR) by exposing Wafer W into ozone atmosphere that the flat nature of the layer insulation film I1 and the property of an MOS transistor are good JAPANESE 74352, A] □ a dinitrogen oxide is used as that the 1st silicon oxide 22 has enough little hygroscopicity and moisture reactant gas, and that a crack does not occur in the 1st silicon oxide 22.

[0138] (b) With the gestalt of the aforementioned implementation, although the silicon oxide using TEOS by plasma CVD was used as the 4th silicon oxide 20, you may use a silicon oxide besides instead of for this. For example, the film formed by the reduced pressure heat CVD using the mono silane and the dinitrogen oxide as such 4th silicon oxide is sufficient. Membranes are faithfully formed to the shape of surface type of a lower layer silicon substrate, and not only coverage nature is good, but since this silicon oxide is precise, even if a passivation function is high and carries out the temperature up of it rapidly in annealing processing further, a crack cannot generate it easily in the 1st silicon oxide 22. Moreover, in order to use heat CVD, there is an advantage without a plasma damage.

[0139] However, since the membrane formation by this method needs to set wafer temperature as about 750-800 degrees C, when using the film which is easy to oxidize like titanium silicide as Salicide structure, it cannot be applied, but it needs to use tungsten silicide or molybdenum silicide.

[0140] (c) With the gestalt of the aforementioned implementation, although the 1st layer insulation film I1 consists of silicon oxides of four layers, it may add not only this but other silicon oxides. For example, you may form the PSG film (concentration of Lynn); 1 - 6 % of the weight) of 100-300nm of thickness formed by the plasma CVD method between the 4th silicon oxide 20 and the 1st silicon oxide 22. By putting in this PSG film, it was checked that the gettering function of a movable ion improves further and the threshold property of a transistor and change of the quiescent current decrease.

[0141] (d) With the gestalt of the above-mentioned implementation, in the layer insulation films I1 and I2, the 3rd silicon oxide 26 and 76 was formed and flattening of this was further carried out by CMP. However, since it has the flat nature excellent in itself, the 1st silicon oxide 22 and 72 does not necessarily need to form these silicon oxides 26 and 76.

[0142] In addition, although the gestalt of the above-mentioned implementation described the semiconductor device including a two-layer wiring field, this invention is applicable to the semiconductor device containing various kinds of elements, such as not only the semiconductor device that can apply also to the semiconductor device which, of course, includes the wiring field of three or more layers, and contains the N channel type MOS device but a P channel type or a CMOS type element.

[0143]

[Translation done.]

ABSTRACTED-PUB-NO: JP 11074352A

BASIC-ABSTRACT:

NOVELTY - A silicon oxide film (22) is formed on a silicon substrate, by polycondensation reaction of silicon and hydrogen peroxide. Another silicon oxide film (24) doped with phosphorus is formed on the film (22). A through-hole (32) is formed on two silicon oxide films, on which a barrier layer (33) is formed. A metal wiring layer (30) is formed over the barrier layer.

DETAILED DESCRIPTION - An **INDEPENDENT CLAIM** is included for semiconductor device manufacturing method.

USE - In LSI manufacture.

ADVANTAGE - Enables formation of flat insulation film at low temperature. Enables miniaturization of semiconductor device,

below half micron and forms reliable contact structure. **DESCRIPTION OF DRAWING(S)** - The figure shows cross-sectional view of layer insulation film arrangement in semiconductor device. (22,24) Silicon oxide films; (32) Through-hole; (33) Barrier layer.

ABSTRACTED-PUB-NO: US 6137176A

EQUIVALENT-ABSTRACTS:

NOVELTY - A silicon oxide film (22) is formed on a silicon substrate, by polycondensation reaction of silicon and hydrogen peroxide. Another silicon oxide film (24) doped with phosphorus is formed on the film (22). A through-hole (32) is formed on two silicon oxide films, on which a barrier layer (33) is formed. A metal wiring layer (30) is formed over the barrier layer.

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ADVANTAGE - Enables formation of flat insulation film at low temperature.

Enables miniaturization of semiconductor device,

below half micron and forms reliable contact structure. DESCRIPTION OF DRAWING(S) - The figure shows cross-sectional view of layer insulation film arrangement in semiconductor device. (22,24) Silicon oxide films; (32) Through-hole; (33) Barrier layer.

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NOVELTY - A silicon oxide film (22) is formed on a silicon substrate, by polycondensation reaction of silicon and hydrogen peroxide. Another silicon oxide film (24) doped with phosphorus is formed on the film (22). A through-hole (32) is formed on two silicon oxide films, on which a barrier layer (33) is formed. A metal wiring layer (30) is formed over the barrier layer.

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CHOSEN-DRAWING: Dwg.2/7

TITLE-TERMS: LAYER INSULATE FILM ARRANGE LSI MANUFACTURE THROUGH HOLE BARRIER
LAYER METAL WIRE LAYER FORMING SEQUENCE SILICON OXIDE FILM TURN
FORMING SILICON SUBSTRATE

DERWENT-CLASS: L03 U11

CPI-CODES: L04-C12A; L04-C13A;

EPI-CODES: U11-C05B1; U11-C05B7;

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